## Low Dimensional Systems and Nanostructures

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- 1. Length scales and low dimensional systems
- 2. Electronic states in confined systems and low dimensions
- 3. Response properties of the electron system in reduced dimensions

Outline

- 1.1 Nano, meso, and macro scales
- 1.2 Dimensionality
  - Chemical bonding approach
  - Physical length scales approach
    - De Broglie length, Fermi wave-length
    - Mean free path
- 1.3 Transport regimes
- 1.4 Examples of low-dimensional systems
  - 2D: transition metal dichalcogenides, semiconducting chalcogenides, layered halogen compounds, graphene, FeSe
  - 1D: polymers, inorganic chains, nanotubes, metallic wires, nanowires on surface
  - 0D: fullerenes, quantum dots, atomic clusters, synthetic nanocrystals

Outline

1.5 Fabrication and characterization techniques

- Nanolitography
- Atomic Force Microscopy (AFM)
- Scanning Tunneling Microscope (STM)
- Molecular Beam Epitaxy (MBE)

1.6 Exercise

1.1 Nano, meso, and macro scales

Macro scale

- The scale of our everyday life
- The properties of materials are defined by physical bulk properties:
  - Color
  - Density
  - Stiffness
  - Sound velocity
  - Bending rigidity
  - ..
- Classical mechanics are enough



1.1 Nano, meso, and macro scales

Meso scale

- The scale in between the bulk and the atomic limits
- Mesoscopic physics study the properties of small condensed objects
- Bulk properties of materials may be used and classical mechanics may be used
- Quantum mechanical effects appear and may be important



1.1 Nano, meso, and macro scales

#### Nano scale

- The scale of the atomic limit
- Bulk properties make no sense
- The quantum nature of the electrons is crucial
- Quantum mechanical description



#### 1.1 Nano, meso, and macro scales

meso

#### Natural things





#### Man-made things



1.2 Dimensionality

#### Chemical bonding approach



ionic bonding electron transferred from Na to Cl



covalent bonding atoms share electrons



metallic bonding ions surrounded by free electrons



molecular bonding weak electrical attraction binds molecules

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1.2 Dimensionality

#### Chemical bonding approach

- In a given compound there might be units strongly bonded by covalent bonds
- These units interact among themselves by weak forces, e.g. hydrogen bonds, Van der Waals forces.
- Depending on the dimension of the unit: 0D, 1D, 2D, 3D systems

Chemical bonding approach: 0D

 $P_4Se_3$ 



Chemical bonding approach: 1D



Chemical bonding approach: 2D



Chemical bonding approach: 3D

Diamond



1.2 Dimensionality

#### Van der Waals forces

- Intermolecular forces, or forces between strongly bonded covalent units
  - **Debye forces:** Dipole-dipole interactions
  - **Hydrogen bonds:** Dipoledipole interactions with hydrogen
  - London dispersion forces: Instantaneous dipoleinduced dipole interactions in non-polar materials
  - Dipole-induced dipole interactions

Van der Waals forces

 Debye forces (hydrogen bonds): Dipole-dipole interactions



1.2 Dimensionality

Van der Waals forces

• London dispersion forces: Instantaneous dipole-induced dipole interactions in non-polar materials





Instantaneous uneven distribution of electrons in He atom Instantaneous dipole Induced dipole on neighboring He atom. Resultant attractive force

Van der Waals forces

Dipole-induced dipole
 interactions

F~r<sup>-6</sup>



1.2 Dimensionality

#### Physical length scales approach

- Based on size dependence of a physical property, e.g. electronic or phonon transport
- Reduced dimension if the dimension of the sample is smaller than a characteristic length L<sub>0</sub>



 $L_x, L_y, L_z < L_0$ 

0D: Quantum dot

1D: Quantum wire

2D: Quantum well



1.2 Dimensionality

Physical length scales approach:

• **De Broglie wavelength**: The (wave)length at which a particle with momentum *p* shows wave-like (quantum mechanical) behavior

Fermi wavelength:
 The (wave)length at which an electron in a metal with energy *E<sub>F</sub>* (Fermi energy) shows wave-like (quantum mechanical) behavior

$$\lambda = h/p$$
$$\omega = ck$$

$$\lambda_F = \sqrt{\frac{h^2}{2mE_F}}$$
$$v_F = \frac{h}{\lambda_F m}$$

Physical length scales approach:

Element	E <sub>F</sub> (e∨)	v <sub>F</sub> (10 <sup>6</sup> m/s)	$\lambda_F$ (Å)
Li	4.74	1.29	5.65
Be	7.08	1.28	5.69
К	2.12	0.86	8.47
Pb	9.47	1.83	3.98

1.2 Dimensionality

Physical length scales approach:

• Mean free path:  $L_{m{m}}$ 

The average distance an electron travels before it experiences a scattering process that changes its initial momentum

- Elastic scattering:
  - When the energy of the electron is conserved
  - Impurity scattering mainly
- Inelastic scattering:
  - When the energy of the electron is not conserved
  - Electron-electron and electronphonon scattering mainly

It is related to the relaxation time  $\tau$ , for a material with carrier velocity v.

$$L_m = v\tau$$



1.2 Dimensionality

#### Physical length scales approach:



1.3 Transport regimes

Transport through a constriction:







 $L_m \gg L$ 

1.4 Examples of low-dimensional systems

#### 2D: Transition metal dichalcogenides (TMDs)



## 1. Length scales and low dimensional systems 1.4 Examples of low-dimensional systems

#### 2D: Transition metal dichalcogenides (TMDs)



## 1. Length scales and low dimensional systems 1.4 Examples of low-dimensional systems

#### 2D: Transition metal dichalcogenides (TMDs)



Exist in the 2D limit

#### Novoselov et al., PNAS (2005)

1.4 Examples of low-dimensional systems

#### 2D: Transition metal dichalcogenides (TMDs)



N. Sabari Arul and V. Devaraj Nithya., "Two Dimensional Transition Metal Dichalcogenides" (2019)

1.4 Examples of low-dimensional systems

#### 2D: Transition metal dichalcogenides (TMDs)

Metallic TMDs phase diagrams with charge-density waves (CDWs) and superconductivity (SC)



1.4 Examples of low-dimensional systems

#### 2D: Transition metal dichalcogenides (TMDs)

Similar phase diagram to the high-temperature superconductors



1.4 Examples of low-dimensional systems

#### 2D: Transition metal dichalcogenides (TMDs)

Contradicting results about the CDW temperature in the 2D limit



Xi et al., Nat. Nanotech. (2015)



Ugeda et al., Nat. Phys. (2015)

1.4 Examples of low-dimensional systems

#### 2D: Semiconducting chalcogenides



"Numbering system adopted by the International Union of Pure and Applied Chemistry (UPAC).

D Encyclopædia Britannica, Inc.

1.4 Examples of low-dimensional systems

#### 2D: Semiconducting chalcogenides

Crystal structures and phase diagrams



Rock-salt structure

Distortions of rock-salt structure

1.4 Examples of low-dimensional systems

#### 2D: Semiconducting chalcogenides

Very good thermoelectric materials



1.4 Examples of low-dimensional systems

#### 2D: Semiconducting chalcogenides

Synthesized in the 2D limit and possible ferroelectricity

#### REPORTS

#### FERROELECTRICITY

#### **Discovery of robust in-plane ferroelectricity in atomic-thick SnTe**

Kai Chang,<sup>1,2\*</sup> Junwei Liu,<sup>3,1,2\*</sup> Haicheng Lin,<sup>1,2</sup> Na Wang,<sup>1,2</sup> Kun Zhao,<sup>1,2</sup> Anmin Zhang,<sup>4</sup> Feng Jin,<sup>4</sup> Yong Zhong,<sup>1,2</sup> Xiaopeng Hu,<sup>1,2</sup> Wenhui Duan,<sup>1,2</sup> Qingming Zhang,<sup>4,5</sup> Liang Fu,<sup>3</sup> Qi-Kun Xue,<sup>1,2</sup> Xi Chen,<sup>1,2</sup>† Shuai-Hua Ji<sup>1,2,6</sup>†

Stable ferroelectricity with high transition temperature in nanostructures is needed for miniaturizing ferroelectric devices. Here, we report the discovery of the stable in-plane spontaneous polarization in atomic-thick tin telluride (SnTe), down to a 1–unit cell (UC) limit. The ferroelectric transition temperature  $T_c$  of 1-UC SnTe film is greatly enhanced from the bulk value of 98 kelvin and reaches as high as 270 kelvin. Moreover, 2- to 4-UC SnTe films show robust ferroelectricity at room temperature. The interplay between semiconducting properties and ferroelectricity in this two-dimensional material may enable a wide range of applications in nonvolatile high-density memories, nanosensors, and electronics.

Science (2016)







1. Length scales and low dimensional systems 1.4 Examples of low-dimensional systems

#### 2D: Graphene

A. Geim K. Novoselov





#### Nobel Prize Physics 2010

"for groundbreaking experiments regarding the two-dimensional material graphene "



1.4 Examples of low-dimensional systems

#### 2D: FeSe

- Superconductivity in bulk FeSe at 9K
- On monolayer on top of  $SrTiO_3$  at 65-109K



1.4 Examples of low-dimensional systems

**1D: Polymers and organic molecules** 



1.4 Examples of low-dimensional systems

1D: Inorganic chains

K<sub>2</sub>Pt(CN)<sub>4</sub>Br<sub>0.2</sub> H<sub>2</sub>O



1.4 Examples of low-dimensional systems

1D: Carbon nanotubes



## 1. Length scales and low dimensional systems 1.4 Examples of low-dimensional systems

#### **1D: Metallic wires**

Gold wires produced with STM



Ohnishi et al., Nature (1998)

1.4 Examples of low-dimensional systems

#### 1D: Nanowires on surfaces

Ferromagnetic one-dimensional monatomic metal chains



Gambardella et al., Nature (2002)

1.4 Examples of low-dimensional systems

#### 1D: Overview of some properties

Та	Table 6.3-1. Quasi-one-dimensional materials (SC: superconductorTr:room temperatureGIC:graphite intercalation compound				
co	mpound	peculiarity			
1)	TTF-TCNQ	Peierls transition			
	(TMTSF)2ClO4	$T_{ m c}=1.3~{ m K~SC}$			
	(TMTSF) <sub>2</sub> PF <sub>6</sub>	$T_{\rm c}=0.9~{ m K~SC}$			
		(under pressure)			
2)	KCP	Peierls transition			
	$K_2Pt(CN)_4Br_{0,2} \cdot H_2O$				
3)	$Hg_{2.86}AsF_6$	SC			
4)	$(SN)_x$	${T}_{ m c}=0.33~{ m K}$			
	$(\mathrm{SNBr}_{0,4})_x$	${T}_{ m c}=0.35~{ m K}$			
		3D-SC			
5)	$(CH)_x \cdot SbF_5$	Peierls transition			
6)	$(=C=)_x$	$\alpha$ -Carbyne			
	$(-C=C-)_x$	$\beta$ -Carbyne			
7)	$TaS_3$	Peierls transition			
	NbSe <sub>3</sub>				
8)	graphite	semi-metal			
	$GIC (SbF_5)$	two-dimensional			
	$GIC (AsF_5)$				
	GIC $(C_8K)$	$T_{\rm c} = 0.2 \mathrm{KSC}$			
9)	copper	metal			
	aluminium				
		three-almensional			

G. Lehmann, P. Ziesche, "Electronic properties of metals" 1990

## 1. Length scales and low dimensional systems 1.4 Examples of low-dimensional systems

#### **OD: Fullerenes**



C<sub>60</sub>

1.4 Examples of low-dimensional systems

#### **0D: Quantum dots**

- Semiconducting particles of few nanometers
- Also called artificial atoms
- Size dependent properties



1.4 Examples of low-dimensional systems

#### **OD: Atomic clusters**

 Clusters can be formed when a hot plume of atoms or molecules in a gas are cooled by collision with raregas atoms much as droplets of water are formed when hot steam cools and condenses

1.4 Examples of low-dimensional systems

#### **OD: Atomic clusters**



P. Jena and A. W. Castleman, "Introduction to Atomic Clusters" (2010)

1.4 Examples of low-dimensional systems

#### **OD: Synthetic nanocrystals**

- Chemical synthesis of nanoparticles of CdS, CdSe, CuCl
- Size control (from few nm to 200 nm)



1.5 Fabrication and characterization techniques

#### Nanolitography

- Techniques for etching, writing, printing at the nanoscale
  - Optical litography
  - Electron-beam litography
  - Scanning proble litography
  - Nanoimprint litography

1.5 Fabrication and characterization techniques

Nanolitography



https://www.youtube.com/watch?time\_continue=189&v=PWV9pvdRBNY&feature =emb\_logo 1. Length scales and low dimensional systems 1.5 Fabrication and characterization techniques

#### Atomic Force Microscopy (AFM)

- Detects the fluctuations of the tip induced by the forces of the sample
- Can be used
   for litography



1.5 Fabrication and characterization techniques

#### Scanning Tunneling Microscope (STM)

- Controls the current tunneled from the sample to the atomic tip.
- Can be used to image and manipulate individual atoms.
- Atomic resolution.



1.5 Fabrication and characterization techniques



Perform a literature search and find a material that can be synthesized in low dimensions and or in its bulk form it has low-dimensional features.

- Which are the features in its electronic properties that make it behave as a low-dimensional material?
- Is there any other particular property that makes it behave as a lowdimensional system?